

Analysis Results for Lunar Soil Simulant Using a Portable X-Ray Fluorescence Analyzer

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Acknowledgments

The author gratefully acknowledges the contribution	ons of Therese Howe with KeyMaster Technologies, Inc., Kennewick, WA
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	TRADEMARKS
	report for identification only. This usage does not constitute an official blied, by the National Aeronautics and Space Administration.
	Available from:
NASA Center for AeroSpace Information 7121 Standard Drive Hanover, MD 21076–1320	National Technical Information Service 5285 Port Royal Roa Springfield, VA 2216

301-621-0390

ce ıd Springfield, VA 22161 703–487–4650

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LIST OF ACRONYMS AND SYMBOLS

Al aluminum

Ca calcium

Cr chromium

Fe iron

JSC-1 Johnson Space Center number one

K potassium

Mg magnesium

Mn manganese

MSFC Marshall Space Flight Center

Na sodium

Ni nickel

O oxygen

P phosphorous

PDA personal digital assistant

Si silicon

SiPIN silicon P-type intrinsic N-type detector

Ti titanium

TRACeR taggant recognition and authentication code enabled reader

XRF x-ray fluorescence

TECHNICAL MEMORANDUM

ANALYSIS RESULTS FOR LUNAR SOIL SIMULANT USING THE A PORTABLE X-RAY FLUORESCENCE ANALYZER

1. INTRODUCTION

Long-term habitation missions on the Moon will require that natural lunar resources be used to minimize the amount of material and supplies that must be transported from Earth. For example, lunar soil will potentially be used for oxygen generation, water generation, and as filler for building blocks. NASA's in situ fabrication and repair program is evaluating portable technologies that can assess the chemistry of lunar soil and lunar soil simulants. This Technical Memorandum summarizes the analysis results of Johnson Space Center number one (JSC–1) lunar soil simulant using the taggant recognition and authentication code enabled reader (TRACeRTM) III–IV handheld x-ray fluorescence (XRF) analyzer manufactured by KeyMaster Technologies, Inc. The focus of the evaluation was to determine how well the current instrument configuration would detect and quantify the components of JSC–1.

2. INSTRUMENT DESCRIPTION

The TRACeR III–IV XRF analyzer was selected for evaluation because it provided the capability to detect and quantify elements with relatively low atomic weights including aluminum (Al), titanium (Ti), and silicon (Si) that are significant components of JSC–1. TRACeR is able to analyze for these elements because it generates a vacuum pressure of approximately 1 torr between the detector and the analyzer head by employing a detachable vacuum accessory. Although handheld XRF analyzers were available from several additional manufacturers at the time this study was conducted, they did not have the capability to provide quantification data for Al, Ti, or Si, and therefore were not tested.

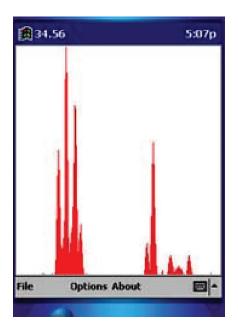
In addition to elemental analysis, the TRACeR III–IV can identify and classify metal alloys. The instrument is programmed with a database of alloy spectra that are mathematically compared to sample spectra to find the best match. Alloy database classifications include iron (Fe), nickel (Ni), and Al. This capability might also prove to be useful during lunar missions if alloy identification is required prior to repair and fabrication operations.

Figure 1 is a photograph of the TRACeR III–IV. The analyzer weighs approximately 4 lb and the analyzer/vacuum pump combination weighs approximately 9 lb. It can be operated using a personal digital assistant (PDA), which is the most portable configuration, and a computer. The instrument uses an x-ray tube as the source and has a Si P-type intrinsic N-type (SiPIN) detector.



Figure 1. TRACeR III–IV portable XRF analyzer.

Figure 2 shows an example of the analysis results as viewed on the PDA screen. An XRF spectrum is provided, along with a breakdown of elemental components and weight percentages.



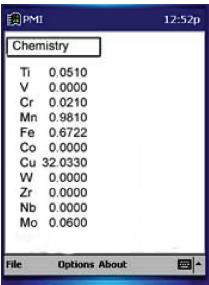


Figure 2. Example of analysis results displayed on the PDA.¹

3. TEST DESCRIPTION

JSC-1 powder was analyzed using a data collection time of 180 s per test and an instrument power level of 15 kV. The simulant was analyzed five times and the results were averaged. The vacuum accessory was employed to enhance detection of Al, Si, and Ti. The instrument was operated using a laptop computer rather than the PDA, because the laptop software provided the operator with more flexibility regarding the selection of elements for quantitative measurements.

A calibration model had to be developed to obtain quantitative JSC-1 chemistry information. Feldspar, an aluminosilicate with the general formula XAI(1-2)Si(2-3)O8, where X is sodium (Na), potassium (K), or calcium (Ca), was used to develop the model. Unfortunately, Feldspar was deficient as a calibration material for several reasons. As shown in table 1, it did not contain Ti, manganese (Mn), chromium (Cr), or phosphorous (P) which are all present in JSC-1. In addition, the Feldspar samples were in solid form while JSC-1 exists as a powder, and the XRF analysis can be impacted by a material's physical characteristics. Even though these deficiencies were recognized, a more representative analog of JSC-1 could not be identified.

Table 1. JSC-1 chemistry.*

Major Components					
Oxide	Concentration (wt%)**	Standard Deviation (wt%)			
SiO ₂	47.71	0.1			
TiO ₂	1.59	0.01			
Al_2O_3	15.02	0.04			
Fe ₂ O ₃	3.44	0.03			
FeO	7.35	0.05			
MgO	9.01	0.09			
CaO	10.42	0.03			
Na ₂ O	2.7	0.03			
K₂0	0.82	0.02			
MnO	0.18	0			
Cr_2O_3	0.04	0			
$P_{2}O_{5}$	0.66	0.01			
2 0	Total = 98.94				
	JSC-1 Trace Elements				
Element	Concentration (ppm)**	Standard Deviation (ppm)			
Scandium	29.2	0.5			
Cobalt	47.7	1.6			
Nickel	137	18			
Rubidium	12.3	1.5			
Cesium	0.339	0.01			
Strontium	860	36			
Barium	822	13			
Lanthanum	48.2	0.9			
Cerium	94.6	1.7			
Neodymium	42	2			
Samarium	7.44	0.13			
Europium	2.18	0.13			
Terbium	0.825	0.04			
Ytterbium	1.99	0.01			
Zirconium	125	3			
Hafnium	3.55	0.08			
Tantalum		0.06			
Uranium	1.96				
Thorium	1.51	0.08			
Arsenic	5.65	0.07			
Selenium	18.7	8.9			
Antimony	<0.5	0			
Tungsten	0.564	0.57			
Gold	36.1	2.6			
Bromine	40.7	29.4			
Lutetium	0.85	0.07			
Lutetium	0.293	0.01			

Source — McKay, D.S.; Carter, J.L.; Boles, W.W.; Allen, C.C.; and Allton, J.H.: "JSC-1: A New Lunar Soil Simulant," Engineering, Construction, and Operations in Space IV, American Society of Civil Engineers, pp. 857–866, 1994.
 ** Wt% and ppm data are the mean of three analyses.

4. TEST RESULTS

A representative XRF spectrum of JSC-1 obtained using the TRACeR III-IV is shown in figure 3, and table 2 summarizes the analysis results. Only the major elemental constituents including Na, Al, Si, K, Ca, Fe, and magnesium (Mg) could be detected and quantified. Mn and Ti were detected but could not be quantified since they were not included in the calibration model. Cr, P, and the numerous JSC-1 trace elements were below the instrument's detection limits.

Standard deviation values for the measured weight percentages were low, which indicated that analysis results were consistent for multiple scans of JSC-1. The percentage error values for chemistry composition were very high, which was due in part to the nonoptimized calibration model.

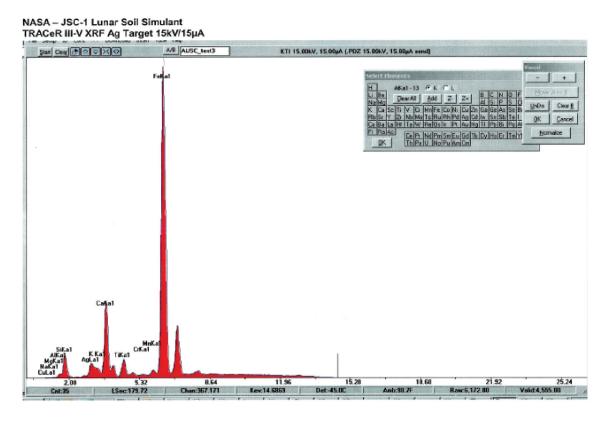


Figure 3. Example XRF spectrum of JSC-1.

Table 2. JSC-1 analysis results using the TRACeR III-IV.

	Measured Wt% of Detected Elements (Present as Oxides)						
Test No.	Na	Mg	Al	Si	K	Ca	Fe
1	1.55	10.03	12.53	34.51	0.1	12.21	16.16
2	1.59	11.15	12.02	34.55	0.17	12.42	15.13
3	1.67	12.87	11.51	34.55	0.12	12.16	15.35
4	1.61	12.47	11.56	34.47	0.06	11	16.12
5	1.59	10.41	12.28	34.49	0.04	11.48	16.16
Average	1.6	11.38	11.98	34.51	0.1	11.85	15.78
Standard Deviation	0.04	1.25	0.44	0.03	0.05	0.59	0.5
Accepted							
Value	2.7	9	15	47	0.8	10.4	10.8
% Error	41%	-26%	20%	27%	88%	-14%	-46%

5. CONCLUSIONS

The TRACeR III–IV portable XRF analyzer was only able to detect and quantify constituents of JSC–1 that were present at weight levels of approximately 1–2 percent and higher. KeyMaster, the manufacturer, advised that quantification of Mn and Ti would likely be possible if they were included in the calibration model. However, responses for Cr, P, and the trace elements would likely remain below detection limits even with enhancement of the model. The instrument was stable in that repeated JSC–1 analysis provided consistent quantitative chemistry values.

Three significant efforts would be required to bring the TRACeR III–IV to a status more suitable for lunar missions: (1) PDA software modification to incorporate the JSC–1 analysis protocol, (2) development of an XRF standard, similar to JSC–1, and (3) refinement of the calibration model.

REFERENCES

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REPORT	DOCUMENTATION	PAGE	Form Approved OMB No. 0704-0188		
Public reporting burden for this collection of informatio ing the data needed, and completing and reviewing the for reducing this burden, to Washington Headquarters of Management and Budget, Paperwork Reduction Pr	collection of information. Send comments regional Services, Directorate for Information Operation	arding this burden estimate or any other aspect	of this collection of information, including suggestions		
1. AGENCY USE ONLY (Leave Blank)	VERED				
	November 2006	Technical N	Memorandum		
4. TITLE AND SUBTITLE		D . 11 W D	5. FUNDING NUMBERS		
Analysis Results for Luna	ir Soil Simulant Using	a Portable X-Ray			
Fluorescence Analyzer					
6. AUTHORS					
R.E. Boothe					
7. PERFORMING ORGANIZATION NAME(S	S) AND ADDRESS(ES)		8. PERFORMING ORGANIZATION REPORT NUMBER		
George C. Marshall Space	e Flight Center		REPORT NUMBER		
Marshall Space Flight Ce	_		M-1176		
Warshan Space Pright Ce	iller, AL 33012		WI 1170		
9. SPONSORING/MONITORING AGENCY I	JAME(S) AND ADDDESS(ES)		10. SPONSORING/MONITORING		
	, ,		AGENCY REPORT NUMBER		
National Aeronautics and	_		NASA/TM-2006-214709		
Washington, DC 20546–	0001		1415111111 2000 211709		
11. SUPPLEMENTARY NOTES					
Prepared by the Engineer	ing Directorate Materia	ls Test Branch			
1 7 8					
12a. DISTRIBUTION/AVAILABILITY STATE	MENT		12b. DISTRIBUTION CODE		
Unclassified-Unlimited					
Subject Category 35					
Availability: NASA CAS					
13. ABSTRACT (Maximum 200 words)					
Lunar soil will potentially be	a used for evagen generat	ion water generation and	d as filler for building blocks		
during habitation missions		_	_		
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		2	2		
handheld x-ray fluorescence evaluation was to determine					
components of JSC-1.	now wen the current ms	strument configuration w	ould detect and quantity the		
components of JSC-1.					
14. SUBJECT TERMS			15. NUMBER OF PAGES		
Fluorescence, regolith, ch	16				
-, 8, •	16. PRICE CODE				
17. SECURITY CLASSIFICATION OF REPORT	18. SECURITY CLASSIFICATION OF THIS PAGE	19. SECURITY CLASSIFICATION OF ABSTRACT	20. LIMITATION OF ABSTRACT		

Unclassified

Unclassified

Unlimited

Unclassified

National Aeronautics and Space Administration IS20 **George C. Marshall Space Flight Center** Marshall Space Flight Center, Alabama 35812